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Note: The most important thing in whole process is that H + and OH-ions combine to form H 2 O. Reaction of acids with metal Oxides - Metal oxide is a chemical compound that contains oxygen and one metal in its chemical formula. For example, copper oxide (CuO), sodium oxide (Na 2 O), magnesium oxide (MgO). Acids give water along with respective salt when they react with a metal oxide.

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(i) The substance 'X' is calcium oxide (lime). Its chemical formula is CaO. (ii) Calcium oxide reacts vigorously with water to form calcium hydroxide (slaked lime).

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